

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
22 January 2004 (22.01.2004)

PCT

(10) International Publication Number
WO 2004/007782 A1

(51) International Patent Classification⁷: C22C 5/00, 5/04

Duncan, Roy [GB/GB]; 68 Warren Wood Drive, High Wycombe HP11 1EA (GB). HYDE, Robin [GB/GB]; 5 Wilbury Close, Letchworth SG6 4JX (GB).

(21) International Application Number:
PCT/GB2003/003037

(74) Agent: WISHART, Ian, Carmichael: Johnson Matthey Technology Centre, Blounts Court, Sonning Common, Reading RG4 9NH (GB).

(22) International Filing Date: 11 July 2003 (11.07.2003)

(81) Designated States (*national*): JP, KR, US.

(25) Filing Language: English

(84) Designated States (*regional*): European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR).

(26) Publication Language: English

Published:
— with international search report

(30) Priority Data:
0216323.6 13 July 2002 (13.07.2002) GB

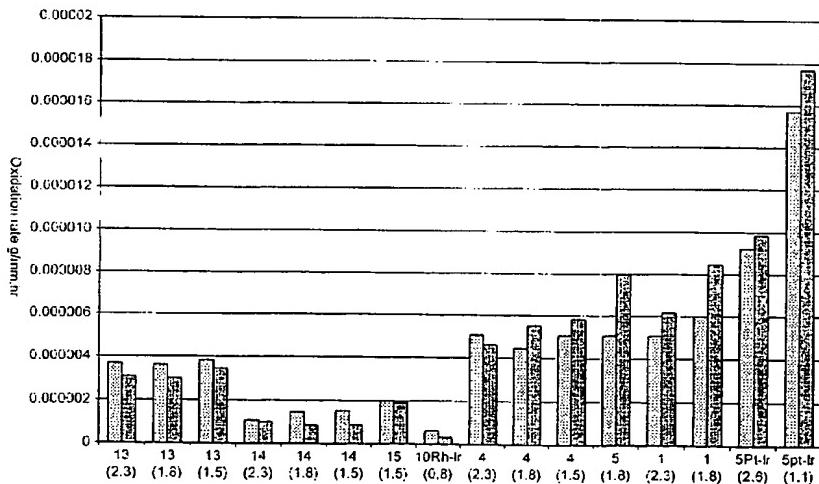
For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(71) Applicant (*for all designated States except US*): JOHN-
SON MATTHEY PUBLIC LIMITED COMPANY
[GB/GB]; 2-4 Cockspur Street, Trafalgar Square, London
SW1Y 5BQ (GB).

(72) Inventors; and

(75) Inventors/Applicants (*for US only*): COUPLAND,

(54) Title: ALLOY



(57) Abstract: An iridium alloy consists essentially of iridium and at least one of W and Zr, and optionally Rh. When present, W comprises between 0.01 and 5 wt % of the alloy; when present in combination with W, Zr comprises between 0.01 and 0.5 wt % of the alloy; when present alone or in combination with Rh only, Zr comprises between 0.01 and 0.09 wt % of the alloy; and when present, Rh comprises between 0.1 and 5 wt % of the alloy. The alloys may be modified by the addition of platinum and other platinum group metals and base metals. The alloys demonstrate enhanced physical and chemical properties and are suitable for use as electrode materials in spark plugs and other high temperature applications.

WO 2004/007782 A1

ALLOY

This invention relates to alloys of iridium, in particular to alloys of iridium with
5 low amounts of alloying elements and uses thereof.

Iridium is a member of the platinum group of metals and has a variety of applications including automobile catalysts, electrodes for industrial electrolysis, crucibles for crystal growth, thermocouples, rocket motor parts, glass making and spark 10 plugs. It has several attractive properties including a very high shear modulus at room temperature and elevated temperature strength second only to tungsten among the refractory metals. It is also thought to be the most corrosion resistant of all metals.

However, despite these benefits there are some disadvantages. Its mechanical 15 properties are sensitive to certain low level impurities and strain rate and it also exhibits a ductile-brittle transition. Due to the rarity of its occurrence in nature its price per gram is of the same order as platinum and furthermore its density is the second highest of all elements. Finally, although compared to the refractory metals its resistance to oxidation 20 is excellent, it nevertheless does exhibit a significant weight loss at elevated temperature under oxidising conditions.

As a result of its scarcity and difficulty in maintaining metal purity during manufacture, the metallurgy of iridium is poorly understood. Indeed little work, relative to that done on alloying of platinum for instance, has been carried out to investigate the 25 effect of alloying on properties. However, the alloying with some elements has been investigated by different workers. Oak Ridge National Laboratories in the USA have been responsible for developing one alloy, DOP-26 based on Ir-0.3W +Th, for radioisotope thermoelectric generator casings used to supply power to spacecraft. Tungsten has been shown to increase the alloy re-crystallisation temperature of iridium 30 by 400°C at >2wt% addition, which makes control of microstructure during hot working, much simpler. Thorium has been shown to promote ductility below the normal ductile/brittle transition zone, although its radioactivity is a major disadvantage when considering this alloy for normal commercial applications. Certain Rare Earth elements, Ce, Y and Lu have also been investigated, and Ce has been found to promote similar

properties to Th, although less pronounced. ORNL have developed a new alloy range based on Ir-0.3 W with low levels of Ce + Th.

US 3,918,965 describes a binary alloy of iridium with 0.3 to 1 wt% hafnium.

5 Improvements in physical properties are claimed.

Work has been limited in respect of alloying iridium with platinum group metals (PGM). Rhodium additions, up to a maximum of ca. 10wt%, have been shown to improve oxidation resistance, ductility and formability. Application of 40%Rh-Ir to 10 novel rocket nozzles was reported in the early 1990's. Ternary alloys have also been long considered for pen nibs, and electrodes. The advent of long life spark plugs has re-invigorated interest in the potential of iridium alloys. Rhodium additions have been found to be beneficial, with 40wt% being best for oxidation resistance. Additions of 10wt% of both platinum and palladium also improve the oxidation resistance of iridium, 15 although not as effectively as rhodium. Al, Si, Cr, Mo and W were found to be ineffective.

EP0866530 A1 discloses ternary and quaternary alloys of iridium, rhodium and at least one of rhenium and ruthenium. Low levels of Re and Ru, either singly or combined, 20 significantly reduce the oxidation loss of an alloy at 1100°C for 30hours, compared to pure iridium. The presence of rhodium is essential, as Re and Ru have little or no effect when combined with iridium alone.

JP 2000290739 A discloses an alloy for the formation of crucibles which can be 25 used at high temperatures without significant deformation or oxidation. The alloy is a binary or ternary alloy of iridium with 0.5-40wt% of Rh and/or Pt.

JP 10259435 A discloses a heat resistant iridium alloy which comprises a base of iridium to which 0.1 to 50wt% of one or more secondary elements is added. Platinum, 30 palladium, rhodium, niobium, tantalum, hafnium, titanium, zirconium, yttrium and lanthanum are suggested as secondary elements however actual examples of only some of these are given, none of which contain secondary elements at less than 1wt%.

US 3,070,450 discloses alloys formed from a base of pure iridium or iridium-0.3wt%W, to which small amounts each of aluminium, iron, nickel, rhodium and thorium are added. The alloys are useful for the encapsulation of radioactive sources so the use of thorium can be tolerated. Thorium containing alloys are not usually suitable
5 for general application.

US 3,293,031 discloses a ductile ternary iridium alloy containing up to 0.5wt% of both titanium and zirconium.

10 Although prior attempts to improve the physical and mechanical properties of iridium by alloying have met with some success, there remains a need for further improvements.

15 In accordance with the present invention, an iridium alloy consists essentially of iridium, at least one of W and Zr and optionally Rh; wherein when present, W comprises between 0.01 and 5 wt% of the alloy; wherein when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; wherein when present alone or in combination with Rh only, Zr comprises between 0.01 and 0.09 wt% of the alloy; and wherein when present, Rh comprises between 0.1 and 5 wt% of the alloy.

20 Preferably, when present, W comprises between 0.01 and 0.5 wt% of the alloy; when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; and when present alone or in combination with Rh only, Zr comprises between 0.02 and 0.07 wt% of the alloy.

25 It will be understood that whilst the amounts of each component are given assuming that the base alloy is pure iridium, in practical terms, the iridium and the alloying elements may contain impurities at levels which would normally be expected for such metals.

30 The alloys of the present invention show enhanced physical and mechanical properties over pure iridium.

The alloy of the present invention may be modified by the addition of Pt in an amount of between 0.1 and 5 wt% of the alloy.

Additionally or alternatively, the alloy of the present invention may be modified
5 by the addition of one or more of Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y, Ti, Ru and Pd individually in an amount of between 0.01 and 10 wt% of the alloy.

Preferably, when present, Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y and Ti individually comprise between 0.01 and 0.5 wt% of the alloy; and when present, Ru and
10 Pd individually comprise between 0.1 and 5 wt% of the alloy.

In a preferred embodiment, the alloy consists essentially of iridium, W and Zr.

In a further preferred embodiment, the alloy consists essentially of iridium and
15 W.

In a yet further preferred embodiment, the alloy consists essentially of iridium and Zr.

20 In measurements of stress rupture times at elevated temperatures, these alloys may outperform pure iridium by a factor of twenty or more. Creep rates at high temperature are also significantly reduced. Furthermore, W and Zr may also retard grain growth at high temperature, with small additions of both W and Zr being found to reduce the rate of grain growth at high temperature by a factor of two compared to pure iridium.

25

In a yet further preferred embodiment, the alloy consists essentially of iridium, Rh, W, and Zr.

30 In a yet further preferred embodiment, the alloy consists essentially of iridium, Pt, Rh, W and Zr.

Significant reduction in weight loss under high temperature oxidising conditions is found for these alloys, when compared to pure iridium.

- In a yet further preferred embodiment, the alloy consists essentially of iridium,
5 Rh and W.

In a yet further preferred embodiment, the alloy consists essentially of iridium,
Rh and Zr.

- 10 In a yet further preferred embodiment, the alloy consists essentially of iridium,
Pt, Rh and W.

In a yet further preferred embodiment, the alloy consists essentially of iridium, Pt
and W. In tensile tests, these alloys demonstrate a considerable increase in elongation to
15 failure compared to pure iridium. In some cases, elongation to failure is increased
two-fold and more.

The enhanced physical and mechanical properties of the alloys of the present
invention make them suitable for use in many high temperature or load bearing
20 applications. For example, they may be used in ignition applications i.e. as components
in spark-plugs or as crucibles, e.g. for crystal growing or other equipment in chemical
and glass applications where high strength, low creep rate and good oxidation resistance
are required. Other applications include electrodes, heat shields and rocket nozzles.
The foregoing examples merely serve to illustrate the many potential uses of the present
25 alloys, and as such, are not intended to be limiting in any way.

The alloys may be manufactured by known methods and fabricated into any
suitable physical form. Improvements in elongation to failure, or ductility, make the
alloys particularly suitable for drawing into wires however, tubes, sheets, grains,
30 powders or other common forms are also contemplated. The alloys may also be used in
spray coating applications.

The invention will now be described by way of example only and with reference to the following drawings in which;

5 Figure 1 is a bar chart comparing the mean elongation at room temperature of an alloy according to the present invention with pure iridium;

Figure 2 is a bar chart comparing the stress rupture time at elevated temperature of four alloys according to the present invention with pure iridium;

10 Figure 3 is a bar chart comparing the rate of grain growth at elevated temperature of four alloys according to the present invention with pure iridium;

Figure 4 is a graph comparing the measured weight loss of two alloys according to the present invention with pure iridium, and;

15 Figure 5 is a bar chart comparing the oxidation rate at two temperatures of several alloys according to the present invention with commercial iridium alloys.

20

EXAMPLE 1
Alloy Preparation

The alloys detailed in table 1 below were prepared by argon arc melting.
25 All values are given in weight percent based on the total weight of the alloy. Balance in all cases is iridium.

Table 1.

Alloy	W	Zr	Rh	Pt	Other
1	0.3	-	-	0.2	-
2	-	0.07	-	-	-
3	0.3	0.02	-	-	-
4	0.05	-	-	-	-
5	0.02	0.02	-	-	-
6	0.3	0.07	2.5	-	-
7	0.3	0.07	2.5	2.5	-
8	0.3	-	2.5	2.5	-
9	0.5	-	1.0	-	-
10	0.3	-	1.0	1.0	-
11	0.3	-	1.0	5.0	-
12	1.0	-	1.0	-	-
13	2.0	-	2.5	-	-
14	0.5	-	2.5	-	-
15	-	0.07	2.5	-	-
16	0.3	-	-	-	-
17	-	-	2.5	-	Ta (0.5)
18	-	-	2.5	-	Nb (0.25)
19	-	-	2.5	-	Mo (0.25)
20	-	-	2.5	-	Cr (0.15)
21	-	-	2.5	-	Pd (0.3)
22	0.05	-	-	5.0	-
23	0.05	-	0.5	5.0	-
24	0.3	-	5.0	1.0	-

EXAMPLE 2
Elongation to Failure

Alloy 1 was hot drawn into wires of 1.8mm diameter, and subjected to tensile
5 testing with a gauge length of 51mm and a cross head speed of 5mm/minute. The result
is shown in Fig. 1. Addition of Pt and W at the ppm level significantly improved the
room temperature mechanical properties of the alloy. Although ultimate tensile strength
was found to only be improved marginally, elongation to failure increased by 117%
relative to similar wires of pure iridium.

10

EXAMPLE 3.
Stress Rupture

Alloys 2-5 were hot rolled into sheets and tensile sample blanks formed by spark
15 erosion machining. These were then surface ground to a thickness of nominally 1.8mm.
The gauge length of each sample blanks was 30mm. Stress rupture times were measured
at a temperature of 1400°C and stress of 75MPa. Results are shown in Fig. 2. Significant
improvements in stress rupture times were found for all alloys compared to pure iridium,
with ppm levels of Zr (alloy 2) or Zr and W (alloy 5) being most effective. Although not
20 shown in Fig.2, creep rates at elevated temperature were also reduced, in some cases by
as much as a factor of 16 compared to pure iridium.

EXAMPLE 4.
Grain Growth Retardation.

25

Alloys 2-5 as detailed in table 1 above, were hot rolled into sheet of nominally
3.5mm thickness . The alloys were held at 1550°C for 400 hours and grain size
measurements made. This was done using an optical microscope. The number of grains
intersecting a line traversing the polished and etched section were counted and averaged
30 over the cross sectional thickness. Results are shown in Fig. 3. Grain growth was
reduced for all alloys compared to pure iridium, with ppm levels of Zr and W (alloy 5)
showing a halving of grain size.

EXAMPLE 5.**Oxidation Weight Loss**

Alloys 6 and 7, as detailed in table 1 above, were hot drawn into wires of
5 between 0.6 and 1.2mm and their weights monitored while being held at 1000°C for
200 hours. Results are shown in Fig. 4. The weight loss of both alloys was
approximately 4 times less than that for pure iridium, over the duration of the test, and
approached that which was found for a commercial 10wt%Rh-Ir alloy.

10 Further oxidation weight loss experiments were carried out using wires of
different thicknesses formed from alloys according to the present invention. Fig. 5 shows
the weight loss rates of alloys 1, 4, 5, 13, 14 and 15. The heavily shaded bars in Fig. 5
represent experiments carried out at 1000°C and the lighter shaded bars represent
experiments carried out at 1100°C. The figure in brackets refers to the thickness of the
15 wire in mm. Oxidation rate is expressed in g/mm.hour. All alloys showed a significant
reduction in oxidation rate compared to a 5%Pt-Ir alloy.

EXAMPLE 6.**Engine Tests**

Alloys 6 and 7, as detailed in table 1 above, were formed into spark plug
electrodes. During testing in a high performance car engine over a period of 175 hours,
the electrodes were found to erode at a similar rate to commercial 10wt%Rh-Ir alloy
25 electrodes, and at a much reduced rate compared to pure iridium electrodes.

CLAIMS

1. An iridium alloy, the alloy consisting essentially of iridium, at least one of W and Zr and optionally Rh; wherein when present, W comprises between 0.01 and 5 wt% of the alloy; wherein when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; wherein when present alone or in combination with Rh only, Zr comprises between 0.01 and 0.09 wt% of the alloy; and wherein when present, Rh comprises between 0.1 and 5 wt% of the alloy.
- 10 2. An alloy according to claim 1, wherein when present, W comprises between 0.01 and 0.5 wt% of the alloy; wherein when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; wherein when present alone or in combination with Rh only, Zr comprises between 0.02 and 0.07 wt% of the alloy; and wherein when present, Rh comprises between 0.1 and 5 wt% of the alloy.
- 15 3. An iridium alloy, the alloy consisting essentially of an alloy according to claim 1 or claim 2, modified by the addition of Pt in an amount of between 0.1 and 5 wt% of the alloy.
- 20 4. An iridium alloy, the alloy consisting essentially of an alloy according to any of claims 1 to 3, modified by the addition of one or more of Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y, Ti, Ru and Pd individually in an amount of between 0.01 and 10 wt% of the alloy.
- 25 5. An alloy according to claim 4, wherein when present, Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y and Ti individually comprise between 0.01 and 0.5 wt% of the alloy; and wherein when present, Ru and Pd individually comprise between 0.1 and 5 wt% of the alloy.
- 30 6. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, W and Zr.

7. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium and W.
8. An alloy according to claim 1 or claim 2, the alloy consisting essentially of
5 iridium and Zr.
9. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh, W, and Zr.
10. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt, Rh,
W and Zr.
15
11. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh and W.
12. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh and Zr.
13. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt, Rh
20 and W.
14. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt and W.
- 25 15. An iridium alloy, the alloy consisting essentially of iridium, Rh and at least one of Ta, Nb, Mo, Cr and Pd; wherein Rh comprises between 0.1 and 5 wt% of the alloy; and wherein Ta, Nb, Mo, Cr and Pd individually comprise between 0.1 and 10 wt% of the alloy.
- 30 16. An electrode comprising an alloy according to any preceding claim.
17. A spark plug comprising an electrode according to claim 16.

18. The use of an alloy according to any of claims 1 to 15 in a high temperature application.

19. The use of an alloy according to any of claims 1 to 15 in a load bearing
5 application.

1 / 3

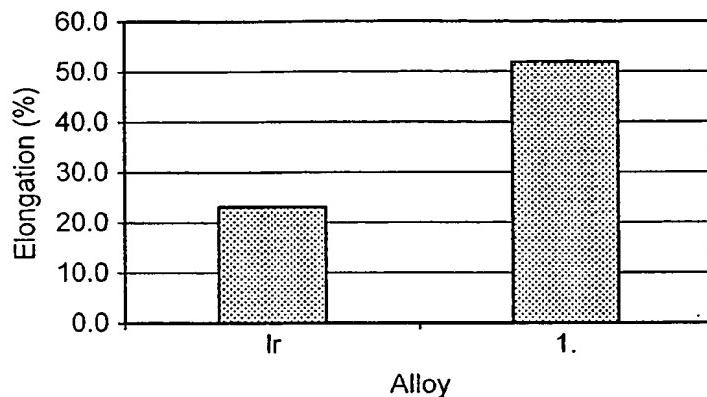


FIG. 1

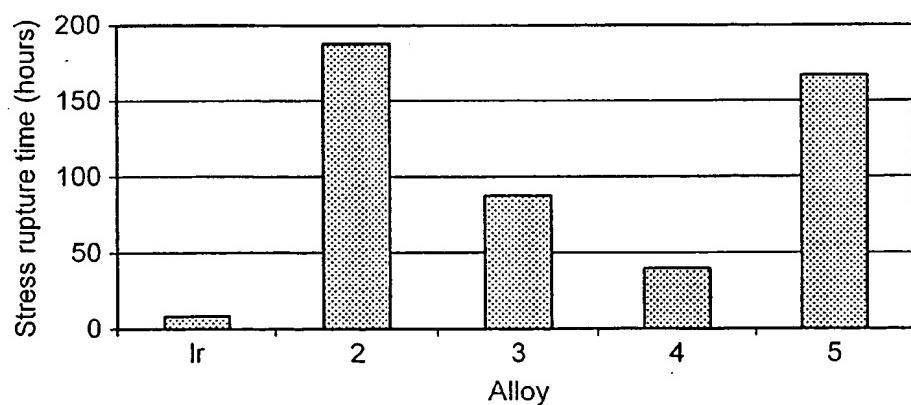


FIG. 2

2 / 3

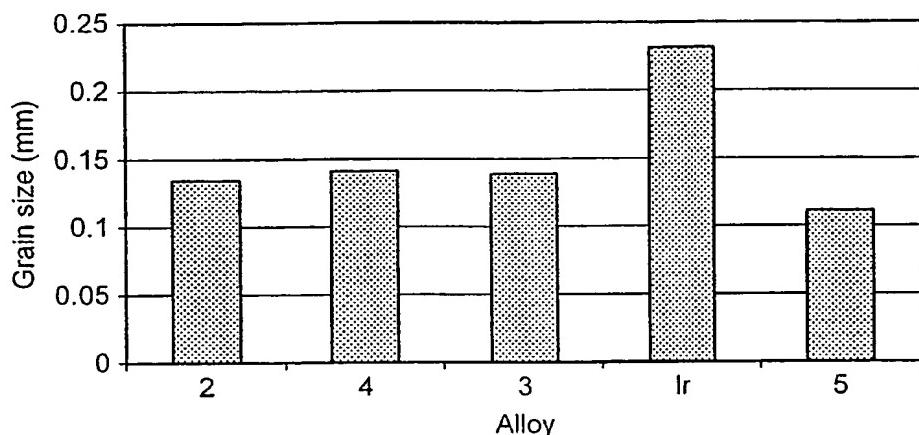


FIG. 3

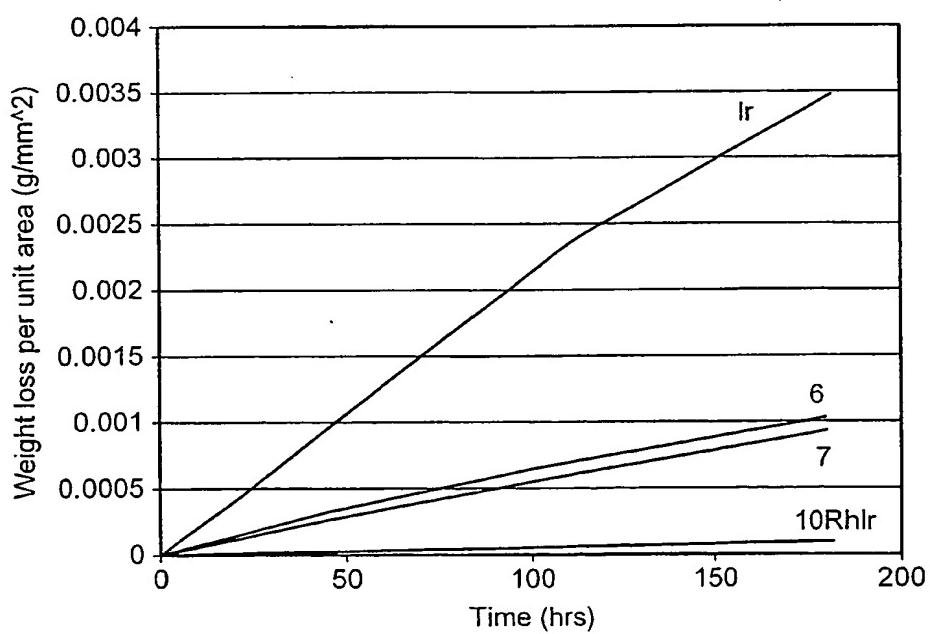


FIG. 4

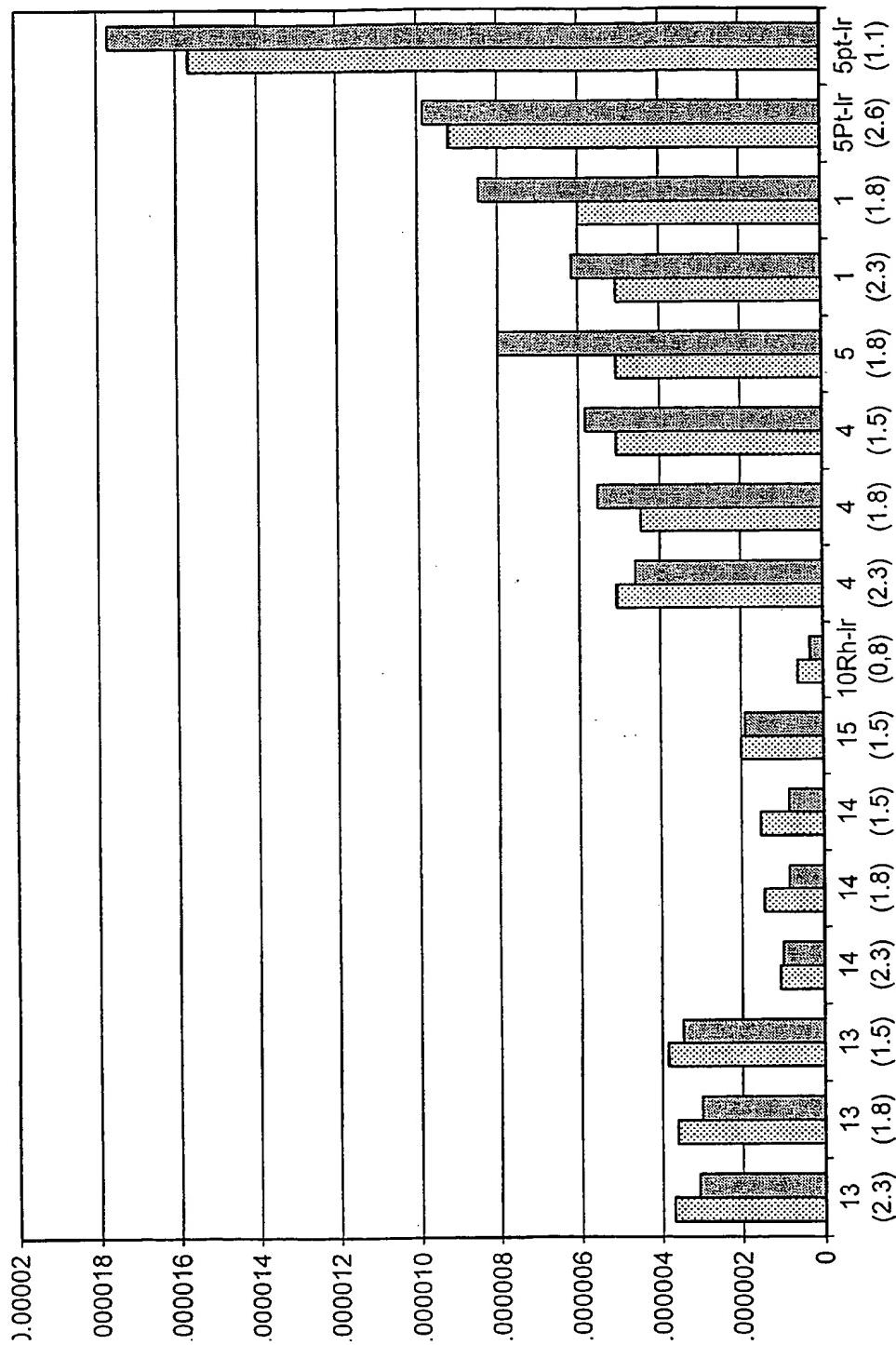


FIG. 5

INTERNATIONAL SEARCH REPORT

International Application No
PCT/GB 03/03037

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C22C5/00 C22C5/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 C22C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 262 779 A (DEZSO ROMHANYI LASZLO ET AL) 26 July 1966 (1966-07-26) claim 1	1-19
X	US 3 293 031 A (CRESSWELL PETER C ET AL) 20 December 1966 (1966-12-20) cited in the application claim 1	1-19
X	GB 1 051 224 A (INTERNATIONAL NICKEL LTD) 14 December 1966 (1966-12-14) claim 1	1-19

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

Z document member of the same patent family

Date of the actual completion of the international search	Date of mailing of the international search report
7 October 2003	14/10/2003

Name and mailing address of the ISA
European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel (+31-70) 340-2040, Tx. 31 651 epo nl.
Fax: (+31-70) 340-3016

Authorized officer

Gregg, N

INTERNATIONAL SEARCH REPORT

International Application No
PCT/GB 03/03037

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT		
Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	PATENT ABSTRACTS OF JAPAN vol. 1998, no. 11, 30 September 1998 (1998-09-30) -& JP 10 162931 A (NGK SPARK PLUG CO LTD), 19 June 1998 (1998-06-19) abstract -----	1-19
X	HUANG C ET AL: "IR-BASED REFRACTORY SUPERALLOYS BY PULSE ELECTRIC CURRENT SINTERING (PECS) PROCESS (II PREALLOYED POWDER)" JOURNAL OF MATERIALS ENGINEERING AND PERFORMANCE, ASM INTERNATIONAL, MATERIALS PARK, US, vol. 11, no. 1, February 2002 (2002-02), pages 32-36, XP001089747 ISSN: 1059-9495 page 32, column 2, paragraph 3 -----	1-19
X	PATENT ABSTRACTS OF JAPAN vol. 013, no. 362 (C-625), 14 August 1989 (1989-08-14) -& JP 01 119595 A (TANAKA KIKINZOKU KOGYO KK), 11 May 1989 (1989-05-11) abstract -----	1-19
X	WO 00 21110 A (CHOI JONG SEO ; JOO KYU NAM (KR); KIM YOON CHANG (KR); SAMSUNG SDI) 13 April 2000 (2000-04-13) claim 1; table 1 -----	1

INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 03/03037

A. CLASSIFICATION OF SUBJECT MATTER
IPC 7 C22C5/00 C22C5/04

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 7 C22C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, PAJ, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3 262 779 A (DEZSO ROMHANYI LASZLO ET AL) 26 July 1966 (1966-07-26) claim 1 ---	1-19
X	US 3 293 031 A (CRESSWELL PETER C ET AL) 20 December 1966 (1966-12-20) cited in the application claim 1 ---	1-19
X	GB 1 051 224 A (INTERNATIONAL NICKEL LTD) 14 December 1966 (1966-12-14) claim 1 ---	1-19

Further documents are listed in the continuation of box C.

Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the International filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

- *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- *X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- *Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- *&* document member of the same patent family

Date of the actual completion of the international search

7 October 2003

Date of mailing of the international search report

14/10/2003

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Gregg, N

INTERNATIONAL SEARCH REPORT

International Application No

PCT/GB 03/03037

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	PATENT ABSTRACTS OF JAPAN vol. 1998, no. 11, 30 September 1998 (1998-09-30) -& JP 10 162931 A (NGK SPARK PLUG CO LTD), 19 June 1998 (1998-06-19) abstract ---	1-19
X	HUANG C ET AL: "IR-BASED REFRACTORY SUPERALLOYS BY PULSE ELECTRIC CURRENT SINTERING (PECS) PROCESS (II PREALLOYED POWDER)" JOURNAL OF MATERIALS ENGINEERING AND PERFORMANCE, ASM INTERNATIONAL, MATERIALS PARK, US, vol. 11, no. 1, February 2002 (2002-02), pages 32-36, XP001089747 ISSN: 1059-9495 page 32, column 2, paragraph 3 ---	1-19
X	PATENT ABSTRACTS OF JAPAN vol. 013, no. 362 (C-625), 14 August 1989 (1989-08-14) -& JP 01 119595 A (TANAKA KIKINZOKU KOGYO KK), 11 May 1989 (1989-05-11) abstract ---	1-19
X	WO 00 21110 A (CHOI JONG SEO ; JOO KYU NAM (KR); KIM YOON CHANG (KR); SAMSUNG SDI) 13 April 2000 (2000-04-13) claim 1; table 1 -----	1

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 03/03037

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
US 3262779	A	26-07-1966	GB CH DE NL	974057 A 411472 A 1224936 B 300224 A	04-11-1964 15-04-1966 15-09-1966
US 3293031	A	20-12-1966	GB DE FR NL	1016809 A 1257437 B 1421717 A 6414675 A	12-01-1966 28-12-1967 17-12-1965 24-06-1965
GB 1051224	A		DE FR NL	1301584 B 89910 E 6601824 A	21-08-1969 14-12-1967 17-08-1966
JP 10162931	A	19-06-1998	NONE		
JP 01119595	A	11-05-1989	NONE		
WO 0021110	A	13-04-2000	CN EP JP WO KR US	1328693 T 1129463 A1 2002527855 T 0021110 A1 2000028717 A 6511632 B1	26-12-2001 05-09-2001 27-08-2002 13-04-2000 25-05-2000 28-01-2003

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No
PCT/GB 03/03037

Patent document cited in search report	Publication date	Patent family member(s)		Publication date
US 3262779	A 26-07-1966	GB 974057 A		04-11-1964
		CH 411472 A		15-04-1966
		DE 1224936 B		15-09-1966
		NL 300224 A		
US 3293031	A 20-12-1966	GB 1016809 A		12-01-1966
		DE 1257437 B		28-12-1967
		FR 1421717 A		17-12-1965
		NL 6414675 A		24-06-1965
GB 1051224	A	DE 1301584 B		21-08-1969
		FR 89910 E		14-12-1967
		NL 6601824 A		17-08-1966
JP 10162931	A 19-06-1998	NONE		
JP 01119595	A 11-05-1989	NONE		
WO 0021110	A 13-04-2000	CN 1328693 T		26-12-2001
		EP 1129463 A1		05-09-2001
		JP 2002527855 T		27-08-2002
		WO 0021110 A1		13-04-2000
		KR 2000028717 A		25-05-2000
		US 6511632 B1		28-01-2003

CLAIMS

1. An iridium alloy, the alloy consisting essentially of iridium, at least one of W and Zr and optionally Rh; wherein when present, W comprises between 0.01 and 5 wt% of the alloy; wherein when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; wherein when present alone or in combination with Rh only, Zr comprises between 0.01 and 0.09 wt% of the alloy; and wherein when present, Rh comprises between 0.1 and 5 wt% of the alloy.
- 10 2. An alloy according to claim 1, wherein when present, W comprises between 0.01 and 0.5 wt% of the alloy; wherein when present in combination with W, Zr comprises between 0.01 and 0.5 wt% of the alloy; wherein when present alone or in combination with Rh only, Zr comprises between 0.02 and 0.07 wt% of the alloy; and wherein when present, Rh comprises between 0.1 and 5 wt% of the alloy.
- 15 3. An iridium alloy, the alloy consisting essentially of an alloy according to claim 1 or claim 2, modified by the addition of Pt in an amount of between 0.1 and 5 wt% of the alloy.
- 20 4. An iridium alloy, the alloy consisting essentially of an alloy according to any of claims 1 to 3, modified by the addition of one or more of Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y, Ti, Ru and Pd individually in an amount of between 0.01 and 10 wt% of the alloy.
- 25 5. An alloy according to claim 4, wherein when present, Ta, Nb, Mo, Cr, Ce, Sc, Lu, Co, Ni, Hf, Y and Ti individually comprise between 0.01 and 0.5 wt% of the alloy; and wherein when present, Ru and Pd individually comprise between 0.1 and 5 wt% of the alloy.
- 30 6. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, W and Zr.

7. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium and W.
8. An alloy according to claim 1 or claim 2, the alloy consisting essentially of
5 iridium and Zr.
9. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh, W, and Zr.
- 10 10. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt, Rh, W and Zr.
11. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh and W.
15
12. An alloy according to claim 1 or claim 2, the alloy consisting essentially of iridium, Rh and Zr.
13. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt, Rh
20 and W.
14. An alloy according to claim 3, the alloy consisting essentially of iridium, Pt and W.
- 25 15. An iridium alloy, the alloy consisting essentially of iridium, Rh and at least one of Ta, Nb, Mo, Cr and Pd; wherein Rh comprises between 0.1 and 5 wt% of the alloy; and wherein Ta, Nb, Mo, Cr and Pd individually comprise between 0.1 and 10 wt% of the alloy.
- 30 16. An electrode comprising an alloy according to any preceding claim.
17. A spark plug comprising an electrode according to claim 16.

18. The use of an alloy according to any of claims 1 to 15 in a high temperature application.

19. The use of an alloy according to any of claims 1 to 15 in a load bearing

5 application.